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**Patentanmeldung Nr.    Patent application No.    Demande de brevet n°**

03077645.4

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**R C van Dijk**



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Bezeichnung der Erfindung/Title of the invention/Titre de l'invention:  
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Si aucun titre n'est indiqué se référer à la description.)

Process and apparatus for the polymerization of ethylene

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**TITLE:**

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**Process and apparatus for the polymerization of ethylene**

The present invention relates to a process for the gas-phase polymerization of ethylene, particularly to a gas-phase polymerization process for obtaining high density polyethylene (HDPE) endowed with broad molecular weight distribution, in particular multimodal molecular weight distribution. The obtained polyethylene is particularly suitable for producing items endowed with enhanced stress-crack resistance, such as pipes, blow and injection molded articles.

For polyolefins and for polyethylene in particular, the molecular weight (MW) and the molecular weight distribution (MWD) are fundamental characteristics affecting the physical, and in particular the mechanical properties of the polymer and thus its applications. It is generally recognized in the art that the higher the molecular weight, the higher the mechanical properties. However, polyolefins with a high molecular weight can be difficult to process, due to their poor flowability properties. The latter can be a serious disadvantage in all those applications requiring a rapid transformation at high shear-rate, for example in blowing and extrusion techniques. In order to improve the rheological properties, while maintaining superior mechanical properties of the final products, it is known in the art to broaden the molecular weight distribution of the polyethylene: the high molecular weight (HMW) fraction contributes to enhance the mechanical properties, the low molecular weight (LMW) fraction helps the processing of the polymer.

The molecular weight distribution can be completely defined by means of a curve obtained by gel permeation chromatography (GPC). Generally, the MWD is defined by a parameter, known as the dispersion index  $D$ , which is the ratio between the average molecular weight by weight ( $M_w$ ) and the average molecular weight by number ( $M_n$ ). The dispersion index constitutes a measure of the width of the MWD. For most applications, the dispersion index varies between 10 and 30.

Another parameter commonly used to define the molecular weight distribution is the ratio between melt index values obtained in different conditions. For instance, with relevance to HDPE for pipe or film application, a ratio between the melt index  $F$  (ASTM-D 1238, 190°C/21.6 Kg) and the melt index  $P$  (ASTM-D 1238, 190°C/5 Kg) higher than 22 is indicative of a significant broad molecular weight distribution.

Higher toughness, strength and stress-crack resistance are required for many high den-

sity polyethylene (HDPE) applications. In addition to these superior mechanical properties, it is important to keep under control production costs, by limiting the use of energy and by increasing the processing yields. High molecular weight HDPE having bimodal or multimodal MWD, i.e. the polymer has two or more distinct ranges of molecular weight, gives the best answer to the customers demand. This kind of polymers is particularly suitable for producing pipes, films, blow and injection molded articles.

It is well known in the art that an insuperable problem of non-homogeneity occurs when bimodality is obtained simply by melt blending low and high molecular weight ethylene polymers. Thus, other methods have been suggested: two reactors in series, eventually with different catalyst in each reactor, or a single reactor fed with a dual site catalyst. Unfortunately, different catalysts used in two conventional reactors in series still lead to polymers with a lack of homogeneity.

Processes with dual-site catalysts show drawbacks as well: in fact, it is very difficult to control the production split between the relatively low and relatively high molecular weight fractions. Furthermore, different catalysts are necessary for obtaining different products, so that a very low operational flexibility can be assured.

Chromium catalysts tend to broaden the MWD of polyolefins, and in some cases can produce bimodal distribution, but the low molecular weight fraction contains a minor amount of comonomer, which depresses the overall mechanical properties, and the stress-crack resistance in particular.

By using two reactors in series with Ziegler/Natta catalyst systems and tailoring the process conditions, it is possible to produce a wide range of high density polyethylene having a large MWD, and in particular a bimodal MWD. In fact, each reactor can work at different polymerization conditions, in terms of catalyst, pressure, temperature, monomer(s) and molecular weight regulator(s) concentration.

USP 6,221,982 discloses a process for producing HDPE in the presence of a Ziegler/Natta catalyst system in two liquid full loop reactors in series. In the first reactor ethylene is homopolymerized or copolymerized with an  $\alpha$ -olefinic monomer comprising from 3 to 8 carbon atoms and, in a second reactor serially connected to the first reactor the product of the first reactor is copolymerized from ethylene and a  $C_3$ - $C_8$   $\alpha$ -olefinic comonomer. The process further requires adding a dehydrogenation step downstream

the first reactor, so that a dehydrogenation catalyst is introduced into the reactants downstream the first reactor.

EP 0 503 791 describes a process for producing bimodal ethylene polymer compositions comprising a mixture of relatively high and low molecular weight polymers by means of two gas-phase, fluidized bed reactors in series. In order to maintain satisfactory processability, the polymer productivity is lowered in the first reaction and raised in the second one. The ethylene partial pressure in each reactor is set accordingly.

However, both the processes of US 6,221,982 and EP 503791 lead to a final polymer having a poor homogeneity. In fact, in each reactor of said cascade-processes a different polymer is generated in term of molecular weight, chemical composition and crystallinity, so that the final polymer shows an intrinsic heterogeneity, caused by the residence time distribution. Depending on the residence time of the polymer particles in the sequence of the reactors, the polymer particles show a larger or a smaller core made of relatively low molecular weight polyethylene and a larger or a smaller outside made of relatively high molecular weight polyethylene (or vice versa). Critical is the homogeneity of the materials for blow-molding products, for films in particular, and for extrusion of pipes, in which the presence of even small quantities of non-homogeneous material brings about the presence of unfused particles (fish-eyes).

An improvement over the prior art is present in the applicant's earlier WO 00/02929 wherein a process for gas-phase polymerization is described. The process is carried out in two interconnected polymerization zones, wherein the growing polymer particles flow through a first polymerization zone (riser) under fast fluidization conditions, leave said riser and enter a second polymerization zone (downcomer) through which they flow in a densified form under the action of gravity, leave said downcomer and are re-introduced into the riser, thus establishing a circulation of polymer between the two polymerization zones. A gas of composition different from that present in the riser is introduced in the downcomer, acting as a barrier to the gas mixture coming from the riser. By properly adjusting the polymerization conditions in said two polymerization zones, it is possible to produce a wide range of bimodal polymers, having an intimate relationship between low and high molecular weight polymer fractions, so that it is possible to achieve a broadening of the MWD, while at the same time increasing the level

of homogeneity of the final polymer.

However, the disclosure of WO 00/02929 does not teach how to obtain high density polyethylen (HDPE) suitable to produce articles having high stress cracking resistance. By way of an example, the proper HDPE for producing pipes is endowed with a high molecular weight and a broad molecular weight distribution, wherein the low molecular weight fraction is an ethylene-homopolymer with high-crystallinity and the high molecular weight fraction is modified with comonomer(s). Therefore, the polymerization should be carried out in such a way to incorporate the comonomer(s) only into the high molecular weight polyethylene fraction. Polymers endowed with such features cannot be obtained according to WO 00/02929, since the comonomer fed to the downcomer, wherein high molecular weight polymer is produced, would inevitably enter the riser, wherein low molecular weight polymer is produced. Consequently, it is not possible to produce in the riser a high crystallinity, low molecular weight homopolymer.

Accordingly, there is a need of improving the polymerization process of WO 00/02929 in order to prepare a large molecular weight distribution HDPE overcoming the prior art drawbacks above discussed.

It has now been found a process for preparing a broad molecular weight polyethylene by polymerizing ethylene in the presence of a polymerization catalyst, the process comprising the following steps, in any mutual order:

- a) polymerizing ethylene, optionally together with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms, in a gas-phase reactor in the presence of hydrogen,
- b) copolymerizing ethylene with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms in another gas-phase reactor in the presence of an amount of hydrogen less than step a),

where in at least one of said gas-phase reactors the growing polymer particles flow upward through a first polymerization zone (riser) under fast fluidization or transport conditions, leave said riser and enter a second polymerization zone (downcomer) through which they flow downward under the action of gravity, leave said downcomer and are reintroduced into the riser, thus establishing a circulation of polymer between said two polymerization zones.

The process according to the present invention allows to obtain from step a) an ethylene polymer with a molecular weight lower than the ethylene copolymer obtained from step b). In particular, the final polymer comprises a high crystallinity, relatively low molecular weight ethylene polymer formed in step a) and intimately mixed with a high molecular weight ethylene copolymer produced in step b). The polymerization process of the present invention allows to bond an increased amount of comonomer only to the relatively high molecular weight polymer fraction thus obtaining ethylene polymers with enhanced mechanical properties and stress-crack resistance, in particular. A further advantage shown by the process of the present invention is that a more effective control of the molecular weight distribution can be achieved.

The stress cracking resistance of the ethylene polymers obtained by the process of the present invention can be evaluated by means of the full notch creep test (FNCT). The full notch creep test (FNCT) is used mainly in Europe by resin producers for development purposes. Depending on the selected test conditions, the rupture time can be strongly reduced with respect to other test methods, such that information can be obtained on highly resistant materials in a short time. The test equipment is simple, being the usual set-up for a tensile creep test. A sample of polymer is immersed in water or a specified surfactant solution at 80°C or 25°C. The sample is notched on four sides perpendicularly to the stress direction and a constant load is applied to the sample. The time to rupture is recorded as a function of the applied stress. The ethylene polymers obtained by the process of the present invention show high values of rupture time, since a high amount of comonomer is bond to the relatively low molecular weight polyethylene fraction.

The above physical-mechanical properties can be obtained carrying out the polymerization of ethylene according to the present invention in two serially interconnected gas-phase reactors. These reactors are described in Patent Application WO 00/02929 and are characterized by two interconnected polymerization zones, in which the polymer particles flow under different fluidization conditions and reactants composition.

In the first polymerization zone (riser), fast fluidization conditions are established by feeding a gas mixture comprising one or more  $\alpha$ -olefins at a velocity higher than the transport velocity of the polymer particles. The velocity of said gas mixture is prefera-



bly comprised between 0.5 and 15 m/s, more preferably between 0.8 and 5 m/s. The terms "transport velocity" and "fast fluidization conditions" are well known in the art; for a definition thereof, see, for example, "D. Geldart, Gas Fluidisation Technology, page 155 et seq., J. Wiley & Sons Ltd., 1986".

In the second polymerization zone (downcomer), the polymer particles flow under the action of gravity in a densified form, so that high values of density of the solid are reached (mass of polymer per volume of reactor), which approach the bulk density of the polymer. In other words, the polymer flows vertically down through the downcomer in a plug flow (packed flow mode), so that only small quantities of gas are entrained between the polymer particles.

A series of two gas-phase reactors with the above configuration is used in the present invention. Preferably, the polymerization of ethylene to produce a relatively low molecular weight ethylene polymer (step a) is performed upstream the copolymerization of ethylene with an  $\alpha$ -olefinic comonomer to produce a relatively high molecular weight ethylene copolymer (step b). To this aim, in step a) a gaseous mixture comprising ethylene, hydrogen and an inert gas is fed to a first gas-phase reactor. The polymerization is carried out in the presence of a highly active catalyst of the Ziegler-Natta or metallocene type. Preferably, no comonomer is fed to said first gas phase reactor and a highly crystalline ethylene homopolymer is obtained in step a). However, a minimal amount of comonomer may be fed with the proviso that the degree of copolymerization in step a) is limited so that the density of the ethylene polymer obtained in step a) is not less than  $0.955 \text{ kg/dm}^3$ , preferably not less than  $0.960 \text{ kg/dm}^3$ , otherwise the distinction between the relatively low molecular weight polymer produced in step a) and the relatively high molecular weight polymer produced in step b) is reduced.

Hydrogen is fed in an amount depending on the catalyst system and, in any case, suitable to obtain in step a) an ethylene polymer with an average molecular weight between 20.000 and 60.000 g/mol and melt flow rate MIE (ASTM D1238, condition 190°C/2.16 Kg) in the range of 10 to 400 g/10 min, preferably 100 to 200 g/10 min. The melt flow rate, commonly referred to as melt index MI, is inversely indicative of the molecular weight of the polymer. In other words, a low melt index is indicative of a high molecular weight for the polyethylene, and vice versa. In order to obtain the above MIE range,

in step a) the hydrogen/ethylene molar ratio is comprised between 0.5 and 5, preferably 1.0 and 3.0, the ethylene monomer being comprised from 5 to 50 % by volume, preferably from 5 to 30 % by volume, based on the total volume of gas present in the first polymerization reactor. The remaining portion of the feeding mixture is represented by inert gases and one or more C<sub>3</sub>-C<sub>12</sub>  $\alpha$ -olefin comonomers, if any. Inert gases which are necessary to dissipate the heat generated by the polymerization reaction are conveniently selected from nitrogen or saturated hydrocarbons, the most preferred being propane.

The operating temperature in the reactor of step a) is selected between 50 and 120 °C, preferably between 80 and 100 °C, while the operating pressure is between 0.5 and 10 MPa, preferably between 2.0 and 3.5 MPa.

According to the present invention, step a) can be also performed in a gas-phase fluidized bed reactor. Nevertheless, both step a) and step b) are preferably carried out in two serially connected gas-phase reactors, in which the growing polymer particles flow upward through a riser under fast fluidization or transport conditions, leave said riser and enter a downcomer through which they flow downward under the action of gravity, leave said downcomer and are reintroduced into the riser, thus establishing a circulation of polymer between said two polymerization zones.

The ethylene polymer obtained in step a) represents from 40 to 65% by weight, preferably from 45 to 55% by weight, of the total ethylene polymer produced in the overall process, i.e. in the first and second serially connected reactors.

The polymer obtained in step a) is discharged from the lower part of the downcomer of the first gas-phase reactor: in said zone the polymer concentration is particularly high, so that only a low amount of gas is discharged together with the polymer.

The ethylene polymer coming from step a) and the entrained gas are then passed through a solid/gas separation step, in order to avoid the gaseous mixture coming from the first polymerization reactor from entering the reactor of step b) (second gas-phase polymerization reactor). Said gaseous mixture can be recycled back to the first polymerization reactor, while the separated ethylene polymer is fed to the reactor of step b). A suitable point of feeding of the polymer into the second reactor is on the connecting part between the downcomer and the riser, wherein the solid concentration is particularly low, so that the flow conditions are not negatively affected.

The operating temperature in step b) is in the range of 65 to 95°C, and the pressure is in the range of 1.5 to 4.0 MPa. The second gas-phase reactor is aimed to produce a relatively high molecular weight ethylene copolymer by copolymerizing ethylene with an  $\alpha$ -olefinic comonomer comprising from 3 to 12 carbon atoms. Furthermore, in order to broaden out the molecular weight distribution of the final ethylene polymer, the reactor of step b) can be conveniently operated by establishing different conditions of monomers and hydrogen concentration within the riser and the downcomer.

To this purpose, in step b) the gas mixture entraining the polymer particles and coming from the riser can be partially or totally prevented from entering the downcomer, so that to obtain two different gas composition zones. This can be achieved by feeding a gas and/or a liquid mixture into the downcomer through a line placed at a suitable point of the downcomer, preferably in the upper part thereof. Said gas and/or liquid mixture should have a suitable composition, different from that of the gas mixture present in the riser. The flow of said gas and/or liquid mixture can be regulated so that an upward flow of gas counter-current to the flow of the polymer particles is generated, particularly at the top thereof, acting as a barrier to the gas mixture entrained among the polymer particles coming from the riser. In particular, it is advantageous to feed a mixture with low content of hydrogen in order to produce the higher molecular weight polymer fraction in the downcomer. One or more comonomers can be fed to the downcomer of step b), optionally together with ethylene, propane or other inert gases. The comonomer may be selected from 1-butene, 1-pentene, 1-hexene, 4-methyl-1-pentene, 1-heptene and 1-octene. Preferably, the comonomer is selected from 1-butene, 1-hexene and 1-octene, more preferably the comonomer is 1-hexene.

The hydrogen/ethylene molar ratio in the downcomer of step b) is comprised between 0.005 and 0.2, the ethylene concentration being comprised from 1 to 20%, preferably 3-10%, by volume, the comonomer concentration being comprised from 0.3 to 5 % by volume, based on the total volume of gas present in said downcomer. The rest is propane or similar inert gases. Since a very low molar concentration of hydrogen is present in the downcomer, by carrying out the process of the present invention is possible to bond a surprisingly high amount of comonomer to the relatively high molecular weight polyethylene fraction.

The polymer particles coming from the downcomer are reintroduced in the riser of step b). Since the polymer particles keep reacting and no more comonomer is fed to the riser, the concentration of said comonomer drops to a range of 0.1 to 3 % by volume, based on the total volume of gas present in said riser. In practice, the comonomer content is controlled in order to obtain the desired density of the final polyethylene. In the riser of step b) the hydrogen/ethylene molar ratio is in the range of 0.05 to 0.3, the ethylene concentration being comprised between 5 and 15 % by volume based on the total volume of gas present in said riser. The rest is propane or other inert gases.

In the reactor of step b) a relatively high molecular weight polymer fraction is produced: the average molecular weight is comprised between 100000 and 1.000.000 g/mol, preferably between 300.000 and 600.000 g/mol.

As above described, two different compositions are present in said reactor of step b), so that it is possible to obtain the relatively high and very-high molecular weight polymer fractions.

The final polymer, discharged through a line placed in the bottom part of the downcomer of the second reactor, is the result of the polymerization in the reactors of step a) and step b). Accordingly, the polymerization process of the invention allows to obtain an ethylene polymer endowed with at least a tri-modal molecular weight distribution: relatively low, high and very-high molecular weights, obtained in the reactor of step a), in the riser of step b) and in the downcomer of step b), respectively.

According to an alternative embodiment of the present invention, it is possible to run the polymerization process so that also the reactor of step a) is operated by establishing different conditions of monomers and hydrogen concentration within the riser and the downcomer. Therefore, it is possible to feed the downcomer of step a) with a gas and/or a liquid having a composition different from that of the gas mixture present in riser. Advantageously, a mixture with a relatively low content of hydrogen can be fed to the upper part of said downcomer, in order to produce an ethylene polymer with an average molecular weight higher than that produced in the riser. In this case, step a) produces a bimodal polyethylene so that the final polymer is endowed with at least a quadrimodal MWD.

The polyethylene obtained by the process of the invention is characterized by a melt in-

dex MIF (ASTM D 1238, condition 190/21,6) in the range of 5 to 40 g/10 min, preferably 10 to 15 g/10 min, and a melt index MIP (ASTM D 1238, 190/5) in the range of 0.1 to 1 g/10 min, preferably 0.15 to 0.6 g/10 min, so that the MIF/MIP ratio is in the range of 20 to 50, preferably 25 to 40. As known, a similar range of MIF/MIP ratio is indicative of a polymer having a broad molecular weight distribution. Typically, the final polyethylene has a high density, comprised between 0.935 and 0.955 kg/dm<sup>3</sup>, preferably between 0.945 and 0.952 kg/dm<sup>3</sup>.

The process of the present invention will now be described in details with reference to the enclosed figure, which is illustrative and not limitative of the scope of the invention. Fig 1 shows a sequence of two gas-phase reactors in which step a), as above defined, is carried out before step b). The first reactor (step a) comprises a riser 1 and a downcomer 2, wherein the polymer particles flow, respectively, upward under fast fluidization condition along the direction of the arrow 14 and downward under the action of gravity along the direction of the arrow 15. The riser 1 and the downcomer 2 are appropriately interconnected by sections 3 and 5. In said first reactor ethylene is polymerized in the presence of hydrogen to produce a relatively low molecular weight ethylene homopolymer. To this aim, a gaseous mixture comprising ethylene, hydrogen and propane is fed to said first reactor through one or more lines 13, suitably placed at any point of the recycling system according to the knowledge of those skilled in art. A mixture of suitable composition comprising ethylene, propane (or other inert) and hydrogen is also fed to the downcomer 2 through one or more lines 18, so that a better control of the reactants composition in said zone can be achieved. The polymerization is carried out in the presence of a highly active catalyst system of the Ziegler-Natta or metallocene type. The various catalyst components are fed through line 12 to the riser 1 at the lower part thereof. After running through the riser 1, the polymer particles and the gaseous mixture leave the riser 1 and are conveyed to a solid/gas separation zone 4. The solid/gas separation can be effected by using conventional separation means such as, for example, a centrifugal separator (cyclone) of the axial, spiral, helical or tangential type.

From the separation zone 4, the polymer enters the downcomer 2. The gaseous mixture leaving the separation zone 4 is recycled to the riser 1 by means of a recycle line 6, equipped with means for the compression 7 and cooling 8.

A part of the gaseous mixture leaving the separation zone 4 can be transferred, after having been compressed and cooled, to the connecting section 5 via the line 9 to favor the transfer of polymer from the downcomer 2 to the riser 1, and to the bottom of the riser 1 through line 10 to establish fast fluidization conditions in the riser 1.

The polymer obtained in step a) is discharged from the lower part of the downcomer 2 and is fed through a line 11 to a solid/gas separator 19, in order to avoid the gaseous mixture coming from the first polymerization reactor from entering the reactor of step b). Said gaseous mixture is fed back to the recycle line 6 through line 20, while the separated ethylene polymer is fed to the second reactor.

The second reactor comprises a riser 1' and a downcomer 2', wherein the polymer particles flow, respectively, upward under fast fluidization conditions along the direction of the arrow 14' and downward under the action of gravity along the direction of the arrow 15'. The two polymerization zones 1' and 2' are appropriately interconnected by section 3' and 5'.

The ethylene polymer exiting the gas/solid separator 19 is fed through line 21 to the connecting section 5' of the second gas-phase reactor.

In said second gas-phase reactor ethylene is copolymerized with 1-hexene in the presence of propane and hydrogen to produce a relatively high molecular weight ethylene copolymer. A gaseous mixture comprising ethylene, hydrogen and propane is fed to said second gas-phase reactor through one or more lines 13', suitable placed at any point of the recycle line 6' according to the knowledge of those skilled in art. Moreover, a gaseous mixture of suitable composition comprising ethylene and 1-hexene is fed to the downcomer 2' through one or more lines 29, so that a better control of the reactants composition in said zone can be achieved.

Analogously to the first reactor, the growing polymer particles and the gaseous mixture leave the riser 1' and are conveyed to a solid/gas separation zone 4'.

From the separation zone 4', the polymer enters the downcomer 2', while the gaseous mixture is collected through line 6', compressed by means of the compression means 7' and split in two. A first part of said mixture is sent to the condenser 22 through line 28, where it is cooled to a temperature at which the monomers and the optional inert gas are partially condensed. The second part of said mixture is cooled by means of the cooling

mean 8' and then fed to the connection zone 5' through line 9' and to the bottom of the riser 1' through line 10'. A separating vessel 24 is placed downstream the condenser 22. The separated gaseous mixture enriched in hydrogen is recirculated through line 26 to the recycle line 6'. On the contrary, the separated liquid is fed to the downcomer 2' through line 27. Said liquid can be fed to said downcomer 2' by gravity by placing the vessel 24 at a convenient height or by any suitable means, such as a pump 25.

The make-up components that must be present in the downcomer 2' in the amounts above stated can be fed in the liquid state directly into the vessel 24 via line 23.

Line 27 for feeding liquid is placed in the upper part of the downcomer 2' and allows to partially or totally prevent the gas mixture coming from the riser 1' from entering the downcomer 2', so as to obtain two different gas composition zones.

The final ethylene polymer resulting from the polymerization of step a) and b) is discharged via line 11'.

The polymerization process of the present invention can be carried out in the presence of a highly active catalyst system of the Ziegler-Natta or metallocene type.

A Ziegler-Natta catalyst system comprises the catalysts obtained by the reaction of a transition metal compound of groups 4 to 10 of the Periodic Table of Elements (new notation) with an organometallic compound of group 1, 2, or 13 of the Periodic Table of element.

In particular, the transition metal compound can be selected among compounds of Ti, V, Zr, Cr, and Hf. Preferred compounds are those of formula  $Ti(OR)_nX_{y-n}$  in which n is comprised between 0 and y; y is the valence of titanium; X is halogen and R is a hydrocarbon group having 1-10 carbon atoms or a COR group. Among them, particularly preferred are titanium compounds having at least one Ti-halogen bond such as titanium tetrahalides or halogenalcoholates. Preferred specific titanium compounds are  $TiCl_3$ ,  $TiCl_4$ ,  $Ti(OBu)_4$ ,  $Ti(OBu)Cl_3$ ,  $Ti(OBu)_2Cl_2$ ,  $Ti(OBu)_3Cl$ .

Preferred organometallic compounds are the organo-Al compounds and in particular Al-alkyl compounds. The alkyl-Al compound is preferably chosen among the trialkyl aluminum compounds such as for example triethylaluminum, triisobutylaluminum, tri-n-butylaluminum, tri-n-hexylaluminum, tri-n-octylaluminum. It is also possible to use

alkylaluminum halides, alkylaluminum hydrides or alkylaluminum sesquichlorides such as  $\text{AlEt}_2\text{Cl}$  and  $\text{Al}_2\text{Et}_3\text{Cl}_3$  optionally in mixture with said trialkyl aluminum compounds. Particularly suitable high yield ZN catalysts are those wherein the titanium compound is supported on magnesium halide in active form which is preferably  $\text{MgCl}_2$  in active form. As internal electron donor compounds can be selected among esters, ethers, amines, and ketones. In particular, the use of compounds belonging to 1,3-diethers, phthalates, benzoates and succinates is preferred.

Further improvements can be obtained by using, in addition to the electron-donor present in the solid component, an electron-donor (external) added to the aluminium alkyl co-catalyst component or to the polymerization reactor. These external electron donor can be the same as, or different from, the internal donor. Preferably they are selected from alkoxysilanes of formula  $\text{R}_a^1\text{R}_b^2\text{Si}(\text{OR}^3)_c$ , where a and b are integer from 0 to 2, c is an integer from 1 to 3 and the sum (a+b+c) is 4;  $\text{R}^1$ ,  $\text{R}^2$ , and  $\text{R}^3$ , are alkyl, cycloalkyl or aryl radicals with 1-18 carbon atoms. Particularly preferred are the silicon compounds in which a is 1, b is 1, c is 2, at least one of  $\text{R}^1$  and  $\text{R}^2$  is selected from branched alkyl, cycloalkyl or aryl groups with 3-10 carbon atoms and  $\text{R}^3$  is a  $\text{C}_1$ - $\text{C}_{10}$  alkyl group, in particular methyl. Examples of such preferred silicon compounds are methylcyclohexyldimethoxysilane, diphenyldimethoxysilane, methyl-t-butyldimethoxysilane, dicyclopentyldimethoxysilane. Moreover, are also preferred the silicon compounds in which a is 0, c is 3,  $\text{R}^2$  is a branched alkyl or cycloalkyl group and  $\text{R}^3$  is methyl. Examples of such preferred silicon compounds are cyclohexyltrimethoxysilane, t-butytrimethoxysilane and hexyltrimethoxysilane.

The above cited catalysts show, in addition to a high polymerization activity, also good morphological properties that make them particularly suitable for the use in the gas-phase polymerization process of the invention.

Also metallocene-based catalyst systems can be used in the process of the present invention and they comprise:

- a) at least a transition metal compound containing at least one  $\pi$  bond;
- b) at least an alumoxane or a compound able to form an alkylmetallocene cation; and
- c) optionally an organo-aluminum compound.

A preferred class of metal compound containing at least one  $\pi$  bond are metallocene compounds belonging to the following formula (I):





wherein

M is a transition metal belonging to group 4, 5 or to the lanthanide or actinide groups of the Periodic Table of the Elements; preferably M is zirconium, titanium or hafnium;

the substituents X, equal to or different from each other, are monoanionic sigma ligands selected from the group consisting of hydrogen, halogen,  $\text{R}^6$ ,  $\text{OR}^6$ ,  $\text{OCOR}^6$ ,  $\text{SR}^6$ ,  $\text{NR}^6_2$  and  $\text{PR}^6_2$ , wherein  $\text{R}^6$  is a hydrocarbon radical containing from 1 to 40 carbon atoms; preferably, the substituents X are selected from the group consisting of -Cl, -Br, -Me, -Et, -n-Bu, -sec-Bu, -Ph, -Bz,  $-\text{CH}_2\text{SiMe}_3$ , -OEt, -OPr, -OBu, -OBz and -NMe<sub>2</sub>;

p is an integer equal to the oxidation state of the metal M minus 2;

n is 0 or 1; when n is 0 the bridge L is not present;

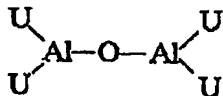
L is a divalent hydrocarbon moiety containing from 1 to 40 carbon atoms, optionally containing up to 5 silicon atoms, bridging Cp and A, preferably L is a divalent group  $(\text{ZR}^7)_2$ , Z being C, Si, and the  $\text{R}^7$  groups, equal to or different from each other, being hydrogen or a hydrocarbon radical containing from 1 to 40 carbon atoms;

more preferably L is selected from  $\text{Si}(\text{CH}_3)_2$ ,  $\text{SiPh}_2$ ,  $\text{SiPhMe}$ ,  $\text{SiMe}(\text{SiMe}_3)$ ,  $\text{CH}_2$ ,  $(\text{CH}_2)_2$ ,  $(\text{CH}_2)_3$  or  $\text{C}(\text{CH}_3)_2$ ;

Cp is a substituted or unsubstituted cyclopentadienyl group, optionally condensed to one or more substituted or unsubstituted, saturated, unsaturated or aromatic rings;

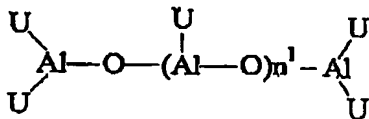
A has the same meaning of Cp or it is a  $\text{NR}^7$ , -O, S, moiety wherein  $\text{R}^7$  is a hydrocarbon radical containing from 1 to 40 carbon atoms;

Alumoxanes used as component b) are considered to be linear, branched or cyclic compounds containing at least one group of the type:



wherein the substituents U, same or different, are defined above.

In particular, alumoxanes of the formula:



can be used in the case of linear compounds, wherein  $n^1$  is 0 or an integer of from 1 to 40 and where the U substituents, same or different, are hydrogen atoms, halogen atoms,  $C_1$ - $C_{20}$ -alkyl,  $C_3$ - $C_{20}$ -cycloalkyl,  $C_6$ - $C_{20}$ -aryl,  $C_7$ - $C_{20}$ -alkylaryl or  $C_7$ - $C_{20}$ -arylalkyl radicals, optionally containing silicon or germanium atoms, with the proviso that at least one U is different from halogen, and j ranges from 0 to 1, being also a non-integer number; or alumoxanes of the formula:



can be used in the case of cyclic compounds, wherein  $n^2$  is an integer from 2 to 40 and the U substituents are defined as above.

The ethylene polymers obtained by the process of the present invention are suitable to prepare a wide range of products, since they achieve an excellent balance of mechanical properties and processing properties. In particular, an excellent level of homogeneity is combined with high values of stress cracking resistance; furthermore, the broadening of the molecular weight distribution helps to achieve good processability and improved flow properties and shear response. In view of these properties, the ethylene polymers obtained by the process of the present invention can be injection or blow-molded into articles, extruded and blown into films or extruded into pipes.

A particularly preferred application is the preparation of pipes able to withstand high pressure. It is conventional to express the performance under stress of polyethylene (or any other thermoplastic) pipes by means of the hoop stress to which a pipe made of polyethylene (or any other thermoplastic) is expected to be able to withstand fifty years at an ambient temperature of 20°C, using water as the test environment (ISO/TR 9080:1992). By the process hereinbefore described, it is possible to obtain PE80 and PE100 pipes, i.e. polyethylene pipes able to withstand fifty years at an ambient temperature of 20°C and a pressure of 8 and 10 MPa, respectively.

The process of the invention will now be described in greater detail with reference to the following examples, being in no way limitative of the object of the invention.

## EXAMPLES

### Characterization

The properties stated were determined according to the following methods:

Melt index E (MIE): ASTM-D 1238, condition 190/2,16

Melt index F (MIF): ASTM-D 1238, condition 190/21,6

Melt index P (MIP): ASTM-D 1238, condition 190/5

Ratio of degrees (F/P): ratio between melt index F and melt index P.

Density: ASTM-D 792.

Flexural elasticity modulus (MEF): the tests were carried out according to ASTM D-790.

Intrinsic viscosity (I.V.): in tetrahydronaphtalene at 135°C.

Stress cracking resistance according to full notch creep test (FNCT): a polymer sample (a small bar 10x10x100 mm), notched on four sides perpendicularly to the stress direction, was immersed in a water solution of *ARCOPAL* (2 % by mole) at 95°C. A constant load of 4.5 MPa was applied to the sample lengthwise to determine the rupture time.

### **Example 1**

The process of the present invention was carried out under continuous conditions in a plant comprising two serially connected gas-phase reactors as shown in Figure 1.

The polymerization was carried out in the presence of a Ziegler-Natta catalyst comprising a solid catalyst component prepared with the procedure described in EP 541 760, Example 1, and triethylaluminium (TEAL) as a cocatalyst. The molar ratio TEAL/Ti is 100.

The catalyst, prepolymerized with propylene, was fed to the first gas-phase polymerization reactor via line 12. In the first reactor ethylene was polymerized with hydrogen in the presence of propane as inert diluent and the amounts of ethylene, hydrogen and propane are specified in Table 1. No comonomer was fed to the first reactor.

Make-up propane, ethylene and hydrogen as molecular weight regulator were fed via line 13. Ethylene and hydrogen were also fed directly into downcomer 2 via line 18.

The properties of the polyethylene resin prepared in the first reactor were analyzed. It can be seen from Table 1 that the polyethylene resin had a melt index MIE of about 120 g/10 min, this representing a relatively low molecular weight for the polymer and a

relatively high density of  $0.968 \text{ kg/dm}^3$ . The first reactor produced around 45 % by weight (split wt %) of the total amount of the final polyethylene resin produced by both first and second reactors. The polymerization was carried out at a temperature of around  $90^\circ\text{C}$  and at a pressure of around 3.0 MPa.

The polymer obtained in the first reactor was continuously discharged via line 11, separated from the gas into the gas/solid separator 19, and reintroduced into the connection section 5' of the second gas-phase reactor via line 21. The second reactor was operated under polymerization conditions at a lower temperature, of about  $75^\circ\text{C}$ , and a lower pressure, of about 2.1 MPa, than those employed in the first reactor.

As a comonomer, 1-hexene was introduced in the second reactor in the amount specified in Table 2. Make-up propane, ethylene and hydrogen were fed through line 13' into the recycling system, while 1-hexene and ethylene were fed directly to the downcomer 2' through line 29.

In order to broaden out the molecular weight distribution of the final ethylene polymer, the second reactor was operated by establishing different conditions of monomers and hydrogen concentration within the riser and the downcomer. This is achieved by feeding via line 27 a liquid stream (liquid barrier) into the upper part of the downcomer 2'. Said liquid stream has a composition different from that of the gas mixture present in the riser. Said different concentrations of monomers and hydrogen within the riser and the downcomer of the second reactor are indicated in Table 2, while the composition of the liquid barrier is indicated in Table 3.

The liquid stream of line 27 comes from the condensation step in the condenser 22, at working conditions of  $50^\circ\text{C}$  and 2.1 MPa, wherein a part of the recycle stream is cooled and partially condensed. The liquid stream 27 can be enriched in 1-hexene by feeding it through line 23.

The final polymer was continuously discharged via line 11'.

The polymerization process in the second reactor produced relatively high molecular weight polyethylene fractions. In Table 5 the properties of the final product are specified. It can be seen that the melt index of the final product is decreased as compared to the ethylene resin produced in the first reactor, showing the formation of high molecular weight fractions in the second reactor. At the same time, the obtained polymer is en-

dowed with a broad molecular weight distribution as witnessed by a ratio MIF/MIP equal to 25.7. The final product was then formed into a small bar (10x10x100 mm) which was subjected to the full notch creep test (FNCT) at a load of 4.5 MPa and a temperature of 95°C. The sample failed the test after about 130 hours, thus showing a high stress cracking resistance.

#### **Examples 2-3**

The process of the invention was carried out with the same apparatus and polymerization catalyst of example 1.

In the first reactor the amounts of ethylene and propane was varied with respect to example 1 and a higher amount of polyethylene resin was produced (split %). The operative conditions in the first reactor are shown in Table 1.

The amount of the 1-hexene comonomer into the downcomer of the second reactor is slightly increased with respect to example 1. The operative conditions in the second reactor are shown in Table 2 and the composition of the barrier stream is shown in Table 3.

The properties of the final polymer are shown in Table 5. The obtained polyethylene resin was formed into a small bar (10x10x100 mm) which was then subjected to the full notch creep test (FNCT) as in example 1. The sample failed the test after about 207 hours (example 2) and about 140 hours (example 3), thus showing a high stress cracking resistance.

#### **Example 4 (comparative)**

An apparatus comprising just one gas-phase polymerization reactor was utilized. Said reactor has the same configuration of the second gas-phase reactor shown in Fig.2.

The same catalyst used in example 1 was fed to the riser of said reactor. The temperature throughout the reactor was kept at about 90°C and the pressure at about 3.0 MPa.

In order to broaden out the molecular weight distribution of the final ethylene polymer, said gas-phase reactor is operated by establishing different conditions of monomers and hydrogen concentration within the riser and the downcomer. This is achieved by feeding a liquid stream (liquid barrier) into in the upper part of the downcomer. The different operative conditions within the riser and the downcomer, and the composition of the liquid barrier are indicated in Table 4.

The properties of the final polymer are shown in Table 5. The obtained polyethylenic resin was formed into a small bar (10x10x100 mm) which was then subjected to the full notch creep test (FNCT) as in example 1. The sample failed the test after only 6,1 hours, thus showing a poor stress cracking resistance.

Table 1 - First Reactor

	Example 1	Example 2	Example 3
<b>Operative conditions</b>			
H <sub>2</sub> /C <sub>2</sub> H <sub>4</sub> Molar Ratio	1.6	1.6	1.6
C <sub>2</sub> H <sub>4</sub> %	12	14	14
C <sub>3</sub> H <sub>8</sub> %	68	63	63
Temp (°C)	90	90	90
Pressure (MPa)	3.0	3.0	3.0
Split (wt %)	45	52	51
<b>Polymer Properties</b>			
MIE (g/10')	120	115	125
Density (kg/dm <sup>3</sup> )	0.9680	0.9678	0.9684
I.V. (dl/g)	0.73	0.75	0.70

Table 2 - Second Reactor

	Example 1	Example 2	Example 3
<b>Operative conditions</b>			
H <sub>2</sub> /C <sub>2</sub> H <sub>4</sub> riser	0.12	0.1	0.12
C <sub>2</sub> H <sub>4</sub> % riser	14	16	14
C <sub>3</sub> H <sub>8</sub> % riser	84	82	84
C <sub>6</sub> H <sub>12</sub> % riser	1.2	1.4	1.1
H <sub>2</sub> /C <sub>2</sub> H <sub>4</sub> downcomer	0.07	0.06	0.07
C <sub>2</sub> H <sub>4</sub> % downcomer	7	8	8
C <sub>3</sub> H <sub>8</sub> % downcomer	90	88	89
C <sub>6</sub> H <sub>12</sub> % downcomer	2	2.5	2.2
Temp (°C)	75	75	75
Pressure (MPa)	2.1	2.1	2.1

Table 3 - Liquid Barrier

	Example 1	Example 2	Example 3
Barrier Feed (kg/h)	200	200	200
C <sub>2</sub> H <sub>4</sub> %	4	4	4
C <sub>3</sub> H <sub>8</sub> %	92	92	92
C <sub>6</sub> H <sub>12</sub> %	3	3.5	3.2
H <sub>2</sub> %	0.04	0.03	0.04

Table 4 - Comparative Example

Operative conditions	
H <sub>2</sub> /C <sub>2</sub> H <sub>4</sub> riser	1.7
C <sub>2</sub> H <sub>4</sub> % riser	16
C <sub>3</sub> H <sub>8</sub> % riser	56
C <sub>6</sub> H <sub>12</sub> % riser	0.3
H <sub>2</sub> /C <sub>2</sub> H <sub>4</sub> downcomer	0.07
C <sub>2</sub> H <sub>4</sub> % downcomer	7
C <sub>3</sub> H <sub>8</sub> % downcomer	91
C <sub>6</sub> H <sub>12</sub> % downcomer	1.1
Temp (°C)	90
Pressure (MPa)	3.0
Liquid barrier	
Barrier Feed (kg/h)	220
C <sub>2</sub> H <sub>4</sub> %	0.2
C <sub>3</sub> H <sub>8</sub> %	98
C <sub>6</sub> H <sub>12</sub> %	1.8
H <sub>2</sub> %	0.005

Table 5 - Final polymer

	Example 1	Example 2	Example 3	Comp. Example
MIP (g/10')	0.3	0.28	0.25	0.32
MIF (g/10')	7.7	8.7	7.6	9
MIF/MIP	25.7	31.1	30.4	28
Density (Kg/dm <sup>3</sup> )	0.9480	0.9477	0.9490	0.9468
I.V. (dl/g)	3.1	3.2	3.3	3
MEF (Mpa)	1042	1032	1080	940
Time for Failure (hrs) - FNCT @ 95°C, 4,5 Mpa	130	207	140	6.1



### CLAIMS

1. A process for preparing a broad molecular weight polyethylene by polymerizing ethylene in the presence of a polymerization catalyst, the process comprising the following steps, in any mutual order:
  - a) polymerizing ethylene, optionally together with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms, in a gas-phase reactor in the presence of hydrogen,
  - b) copolymerizing ethylene with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms in another gas-phase reactor in the presence of an amount of hydrogen less than step a),where in at least one of said gas-phase reactors the growing polymer particles flow upward through a first polymerization zone (riser) under fast fluidization or transport conditions, leave said riser and enter a second polymerization zone (downcomer) through which they flow downward under the action of gravity, leave said downcomer and are reintroduced into the riser, thus establishing a circulation of polymer between said two polymerization zones.
2. The process according to claim 1, wherein from step a) an ethylene polymer is obtained with a molecular weight lower than the ethylene copolymer obtained from step b).
3. The process according to any of claims 1-2, wherein step a) is performed upstream step b).
4. The process according to any of claims 1-3, wherein the density of the ethylene polymer obtained in step a) is not less than  $0.955 \text{ kg/dm}^3$ .
5. The process according to any of claims 1-4, wherein the ethylene polymer obtained in step a) has a melt flow rate MIE in the range of 10 to 400 g/10 min.
6. The process according to claim 5, wherein the MIE is from 100 to 200 g/10 min.
7. The process according to any of claims 1-6, wherein in step a) the hydrogen/ethylene molar ratio is comprised between 0.5 and 5.0, the ethylene monomer being comprised between 5 and 50 % by volume.
8. The process according to any of claims 1-7, wherein the operating temperature in step a) is selected between 50 and 120 °C.

9. The process according to any of claims 1-8, wherein the operating pressure in step a) is between 0.5 and 10 MPa.
10. The process according to claim 1, wherein step a) is performed in a fluidized bed reactor.
11. The process according to claim 1, where in both the gas-phase reactors of step a) and step b) the growing polymer particles flow upward through said riser under fast fluidization conditions, leave said riser and enter said downcomer through which they flow downward under the action of gravity, leave said downcomer and are re-introduced into said riser.
12. The process according to any of claims 1-11, wherein the ethylene polymer obtained in step a) represents from 40 to 65% by weight of the total ethylene polymer produced in the overall process
13. The process according to any of claims 1-12, wherein the ethylene polymer coming from step a) and the entrained gas are passed through a solid/gas separation step and the separated polymer is fed to the reactor of step b).
14. The process according to any of claims 1-13, wherein the operating temperature in step b) is in the range of 65 to 95°C.
15. The process according to any of claims 1-14, wherein the operating pressure in step b) is in the range of 1.5 to 4 MPa.
16. The process according to any of claims 1-15, wherein the reactor of step b) is operated by establishing different conditions of monomers and H<sub>2</sub> concentration within the riser and the downcomer.
17. The process according to claim 16, wherein said different conditions are achieved by feeding a gas and/or a liquid mixture into the downcomer, said gas and/or liquid mixture having a composition different from that of the gas mixture present in the riser.
18. The process according to claim 1, wherein the  $\alpha$ -olefinic comonomer is selected from 1-butene, 1-pentene, 1-hexene, 4-methyl-1-pentene, 1-heptene and 1-octene.
19. The process according to claims 16-18, wherein the hydrogen/ethylene molar ratio in the downcomer of step b) is comprised between 0.005 and 0.2, the ethylene concentration being comprised from 1 to 20 % by volume and the comonomer concen-

- tration from 0.3 to 5 % by volume based on the total volume of gas present in said downcomer.
20. The process according to any of claims 16-19, wherein the hydrogen/ethylene molar ratio in the riser of step b) is comprised between 0.05 and 0.3, the ethylene concentration being comprised from 5 to 15 % by volume and the comonomer concentration from 0.1 to 3.0% by volume based on the total volume of gas present in said riser.
  21. The process according to any of claims 1-20, wherein the ethylene copolymer obtained in step b) has an average molecular weight comprised between 100000 and 1.000.000 g/mol, preferably between 300.000 and 600.000 g/mol.
  22. The process according to claims 1-21, wherein an ethylene polymer endowed with at least a tri-modal molecular weight distribution is obtained.
  23. The process according to claim 22, wherein said ethylene polymer has a melt index MIF in the range of 5 to 40 g/10 min and a melt index MIP in the range of 0.1 to 1 g/10 min.
  24. The process according to claim 23, wherein the MIF/MIP ratio is in the range of 20 to 50.
  25. The process according to any of claims 22-24, wherein said ethylene polymer has a density comprised between 0.935 and 0.955 kg/dm<sup>3</sup>.

**ABSTRACT**

Process for preparing a broad molecular weight polyethylene by polymerizing ethylene in the presence of a polymerization catalyst, the process comprising the following steps, in any mutual order:

- a) polymerizing ethylene, optionally together with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms, in a gas-phase reactor in the presence of hydrogen,
- b) copolymerizing ethylene with one or more  $\alpha$ -olefinic comonomers having from 3 to 12 carbon atoms in another gas-phase reactor in the presence of an amount of hydrogen less than step a),

where in at least one of said gas-phase reactors the growing polymer particles flow upward through a first polymerization zone (riser) under fast fluidization or transport conditions, leave said riser and enter a second polymerization zone (downcomer) through which they flow downward under the action of gravity, leave said downcomer and are reintroduced into the riser, thus establishing a circulation of polymer between said two polymerization zones.

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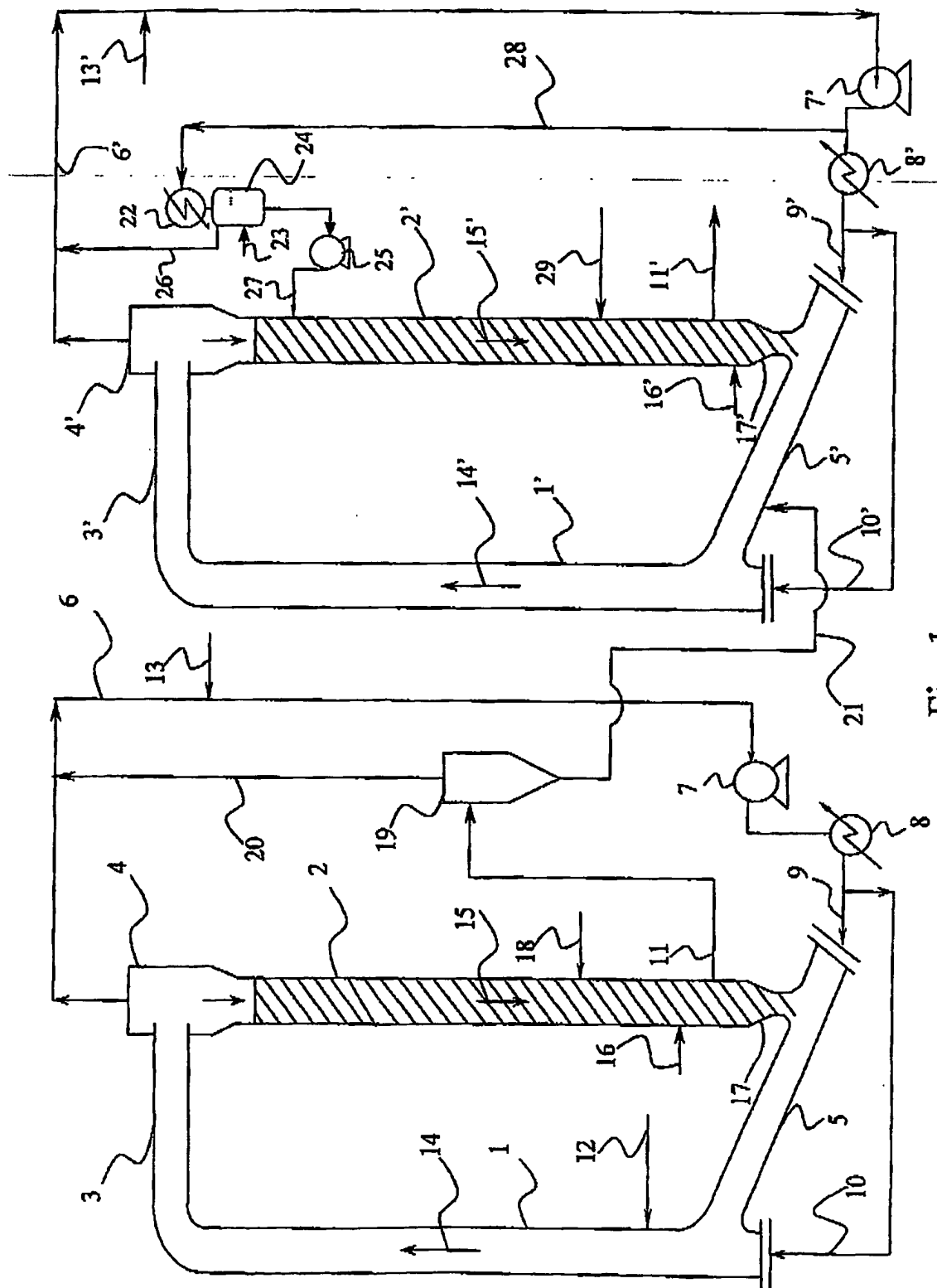


Fig. 1

PCT/EP2004/008063



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